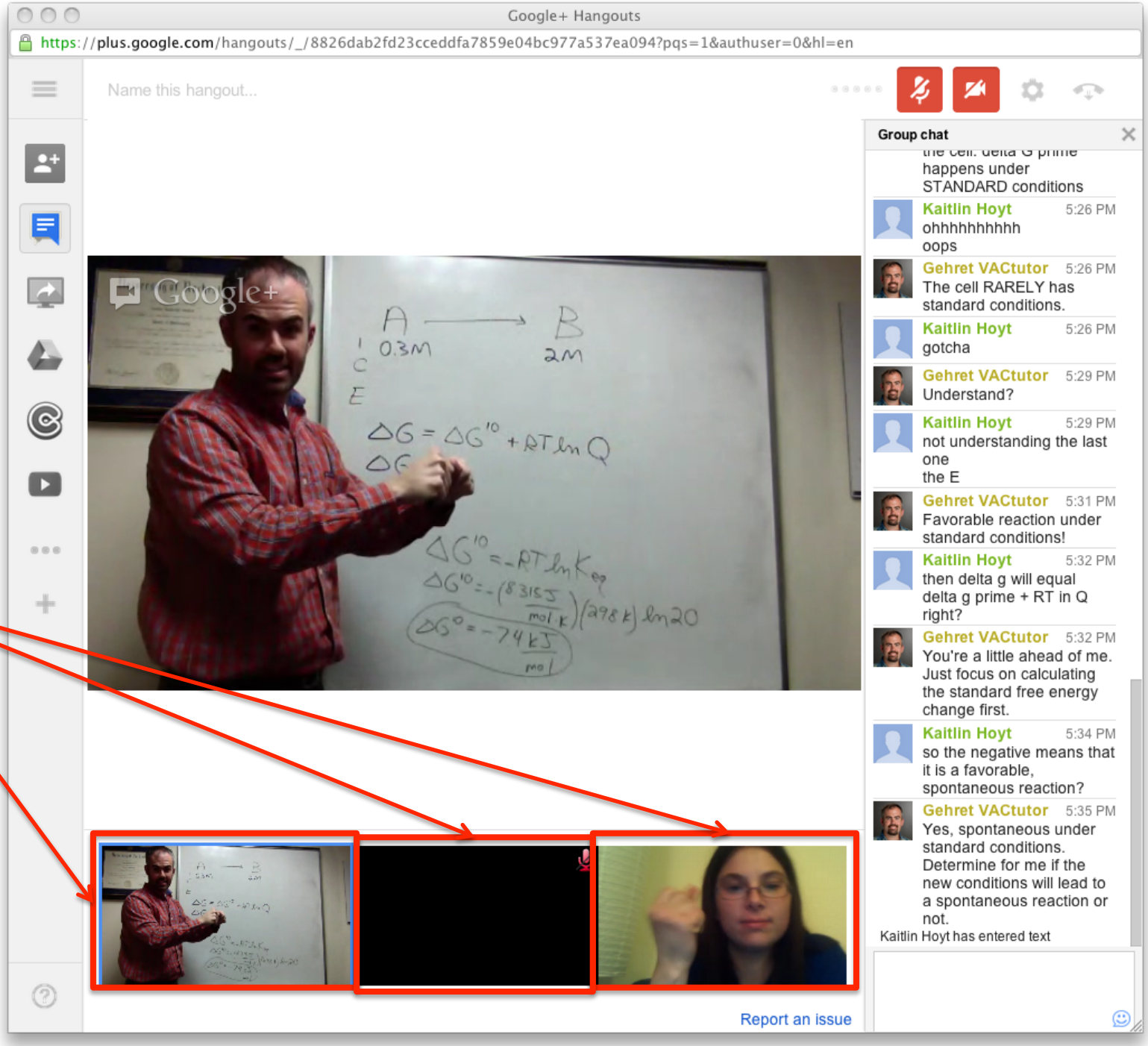
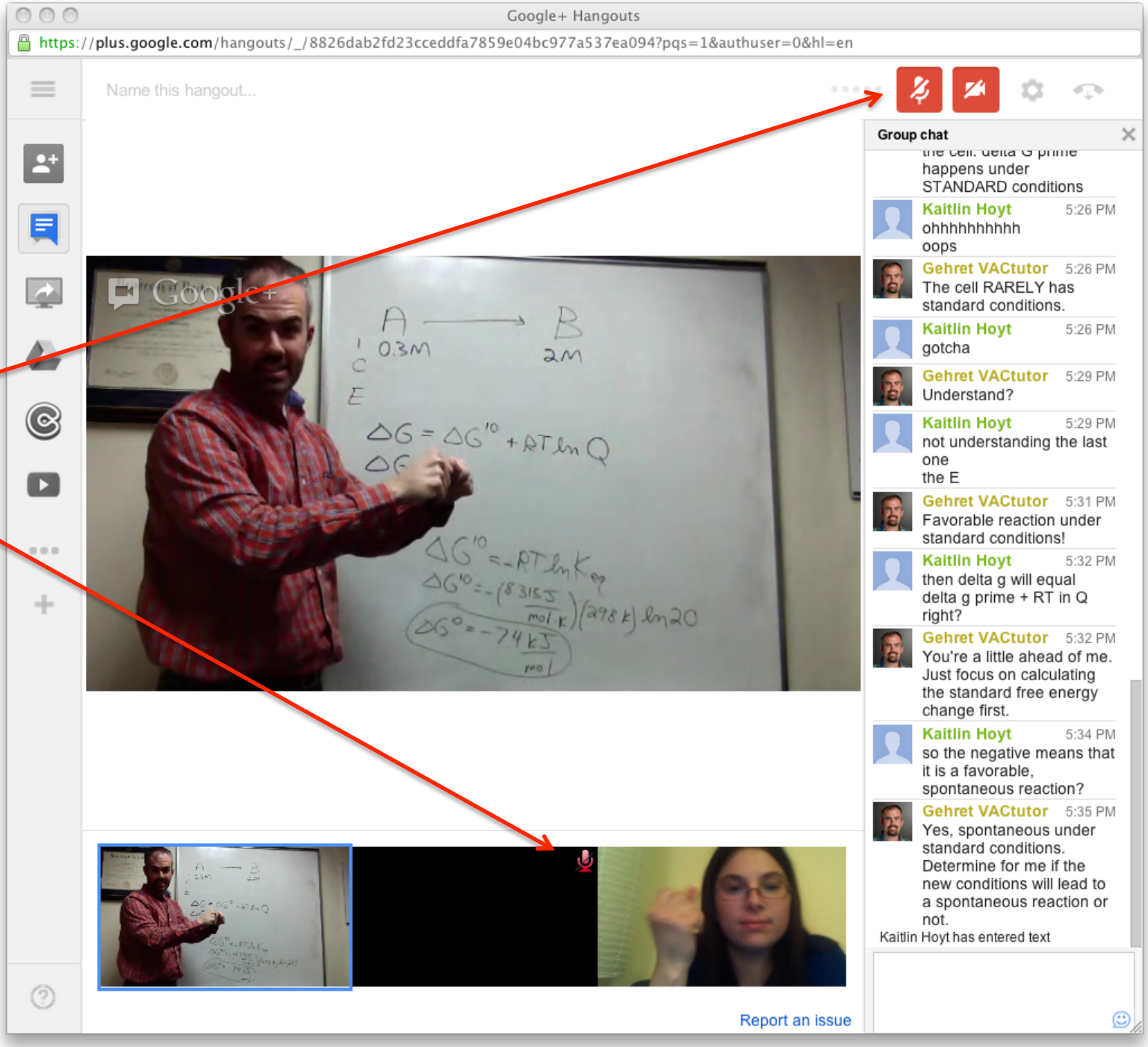
Video Feeds For All Users





Microphone Capability

Controlled Individually

Chat Feature

Google+ Hangouts

https://plus.google.com/hangouts/_/8826dab2fd23cceddfa7859e04bc977a537ea094?pqs=1&authuser=0&hl=en

Name this hangout...

Group chat

happens under STANDARD conditions

Kaitlin Hoyt 5:26 PM
ohhhhhhhhhh
oops

Gehret VACtutor 5:26 PM
The cell RARELY has standard conditions.

Kaitlin Hoyt 5:26 PM
gotcha

Gehret VACtutor 5:29 PM
Understand?

Kaitlin Hoyt 5:29 PM
not understanding the last one
the E

Gehret VACtutor 5:31 PM
Favorable reaction under standard conditions!

Kaitlin Hoyt 5:32 PM
then delta g will equal delta g prime + RT in Q right?

Gehret VACtutor 5:32 PM
You're a little ahead of me. Just focus on calculating the standard free energy change first.

Kaitlin Hoyt 5:34 PM
so the negative means that it is a favorable, spontaneous reaction?

Gehret VACtutor 5:35 PM
Yes, spontaneous under standard conditions. Determine for me if the new conditions will lead to a spontaneous reaction or not.

Kaitlin Hoyt has entered text

Report an issue

Chat Feature can be used to clarify some complex terminology